## THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

[BPHARM 0921]

# SEPTEMBER 2021 (FEBRUARY 2021 EXAM SESSION)

Sub. Code: 4257

# B. PHARMACY DEGREE EXAMINATION SECOND YEAR – (NON-SEMESTER) PAPER II – PHARMACEUTICAL ANALYSIS & PHYSICAL CHEMISTRY *Q.P. Code: 564257*

Time: Three hours Answer ALL questions

# Maximum: 100 Marks

# SECTION-A

# (PHARMACEUTICAL ANALYSIS)

#### I. Elaborate on:

- 1. a) Illustrate briefly on Neutralization curves and the choice of indicators in acid base titrations.
  - b) Explain briefly about Argentometric titrations and its various types.

#### II. Write notes on:

- 1. Write a note on the types of Solvents used in non-aqueous titrations.
- 2. Discuss briefly about Masking and Demasking agent with examples.
- 3. Discuss about Nernst equation.
- 4. Brief out on oxygen flask combustion method.

# III. Short answers on:

- 1. Common ion effect.
- 2. Solubility product.
- 3. Co-precipitation.
- 4. Universal indicators.
- 5. Significant figure.

#### SECTION-B (PHYSICAL CHEMISTRY)

## I. Elaborate on:

1. a) Define Colligative properties. Describe elaborately about the types of Colligative properties.

b) What is Adsorption? Explain briefly about Langmuir adsorption isotherm.

#### II. Write notes on:

- 1. Write a note on Joule Thomson effect.
- 2. Derive Vant-hoff equation and mention its use.
- 3. Enumerate the theory and mechanism of Catalysis.
- 4. Describe briefly theories involved in reaction kinetics.

#### III. Short answers on:

- 1. Second law of Thermodynamics.
- 2. Define the terms phase, component and degree of freedom.
- 3. Bomb calorimeter.
- 4. Factors influencing adsorption.
- 5. Refractive index.

 $(5 \times 2 = 10)$ 

 $(4 \times 5 = 20)$ 

 $(1 \times 20 = 20)$ 

 $(5 \times 2 = 10)$ 

 $(4 \times 5 = 20)$ 

 $(1 \times 20 = 20)$