### THE TAMIL NADU DR. M.G.R. MEDICAL UNIVERSITY

# [BPHARM 1022] OCTOBER 2022 Sub. Code: 2033 (MARCH 2022 EXAM SESSION)

# B.PHARMACY DEGREE COURSE (SEMESTER EXAMINATIONS) PCI Regulation 2017 – SEMESTER III PAPER II - PHYSICAL PHARMACEUTICS I Q.P. Code: 562033

Time: Three hours Maximum: 75 Marks

# I. Elaborate on: Answer any TWO questions.

 $(2 \times 10 = 20)$ 

- 1. What is partially miscible liquids, explain it with neat diagram.
- 2. Define complex. Write in detail about metal ion complex.
- 3. Explain the determination of optical rotation with neat diagram.

## II. Write notes on: Answer any SEVEN questions.

 $(7 \times 5 = 35)$ 

- 1. Explain about the mechanism of solute solvent interactions.
- 2. Dielectric constant.
- 3. Explain Dunouy Ring method with neat diagram.
- 4. Write the different methods by which the pH can be determine.
- 5. Describe Sublimation critical point with neat diagram.
- 6. Write in detail about polymorphism.
- 7. Explain the Job's method of analysis with neat diagram.
- 8. State and derive Raoult's law.
- 9. Thermodynamic treatment of stability constant.

### III. Short answers on: Answer ALL questions.

 $(10 \times 2 = 20)$ 

- 1. Define association.
- 2. Snell's law.
- 3. Define interfacial tension and write its units.
- 4. Write any two examples of complexation.
- 5. Define Sublimation.
- 6. Distribution Law.
- 7. Hypertonic solution.
- 8. Write few applications of buffer.
- 9. Define CMC.
- 10. Detergent.

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