OCTOBER 2015

Sub. Code: 3805

PHARM. D DEGREE EXAMINATION (2009-2010 Regulation) FIRST YEAR PAPER V – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code : 383805

Time : Three hours

I. Elaborate on :

- 1. Define the term Pharmaceutical Aids with suitable examples.
- 2. What is Non Aqueous titration? Explain its principle with suitable example. List out solvent used in the Non-Aqueous titration. How will you prepare and standardize the 0.1N Perchloric acid?
- 3. Explain the theory of precipitation titrations. Describe the principle involved in modified Volhard's method with example.
- 4. What are Electrolytes? Write its essential function in the body. Give note on acidbase regulators.

II. Write notes on :

- 1. What is Ceriometry? Explain its advantages over other oxidizing agents.
- 2. Henderson Hasselbalch equation.
- 3. Limit test for arsenic.
- 4. What is acidifiers and antacid? Give examples.
- 5. Note on Radio Pharmaceuticals.
- 6. Name the titration for the following compounds:

a) Ammonium chloride	b) Copper sulphate	c) Barium sulphate
d) Sodium benzoate	e) Calcium gluconate	

 $(4 \times 10 = 40)$

Maximum: 70 marks

 $(6 \times 5 = 30)$