## FIRST YEAR B.PHARM. EXAM PAPER I – PHARMACEUTICAL INORGANIC CHEMISTRY

Q.P. Code: 564251

Time: Three Hours Maximum: 100 marks

I. Essay: (2X20=40)

- 1. a) Explain about Pauling's Valence bond Theory.
  - b) Discuss about the biological role of co-ordination compounds.
- 2. a) Define Acidifier. Write the method of preparation and assay of Na<sub>2</sub>HPo<sub>4</sub>.
  - b) Write about the combination of Antacid preparation.

II. Short Notes:  $(8 \times 5 = 40)$ 

- 1. Explain Electrolyte combination therapy with examples.
- 2. What are the precautions to be observed while handling radio active materials?
- 3. Write the principle involved in the limit test for Iron.
- 4. Write notes on Dimethicone.
- 5. Write the method of preparation, assay and uses of Chlorinated lime.
- 6. Write about Physiological acid base balance and its importance.
- 7. Define Respiratory stimulant. Write the preparation and assay of any one drug.
- 8. Write notes on antidotes.

III. Short Answers:  $(10 \times 2 = 20)$ 

- 1. Write a note on water for injection.
- 2. Define suspending agent. Give an example.
- 3. What are antimicrobials? Write about their importance.
- 4. Give the uses of Iron dextran injection.
- 5. Explain the importance of test for purity.
- 6. What is the use of Hypertonic saline?
- 7. What are ideal antacid?
- 8. Write the composition and used of Magaldrate.
- 9. Write a note on Povidone Iodine.
- 10. Define valency.

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